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Key indicators

Single-crystal X-ray study

T = 298 K

Mean $\sigma(\text{C}-\text{C}) = 0.003 \text{ \AA}$

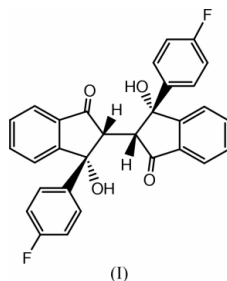
R factor = 0.054

wR factor = 0.190

Data-to-parameter ratio = 16.0

For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.***rac*-3,3'-Bis(4-fluorophenyl)-3,3'-dihydroxy-
2,3,2',3'-tetrahydro-[2,2']biindenyl-1,1'-dione**The title compound, $\text{C}_{30}\text{H}_{20}\text{F}_2\text{O}_4$, has been obtained as a by-product in the preparation of biindenylidene compounds. The molecule has a twofold axis. There are intramolecular O—H···O hydrogen bonds.

Comment

Photochromism of the inclusion crystals of a hydroperoxy-biindenylidene has been reported recently by Tanaka *et al.* (2003). The title compound, (I), was obtained as a by-product in the preparation of the biindenylidene compounds.

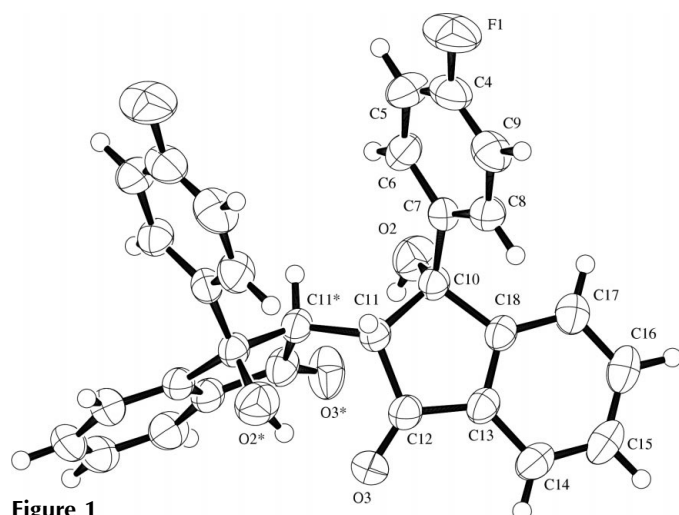
The molecule (I) has a twofold axis, which is parallel to *c* (Figs. 1 and 2). The central C11—C11ⁱ bond [symmetry code: $(i) \frac{1}{2} - x, \frac{1}{2} - y, z$] is a single bond (Table 1), which corresponds to no photochromic property of (I), since the photochromism of these crystals seems to be due to the biradical caused by breaking the π -bond at the center of the biindenylidene moiety (Ohba *et al.*, 2003). The indene ring system is not planar, as the five-membered ring has an envelope conformation, with atom C11 in the flap position. The angle between the planes composed of atoms C12—C18/C10 and C10—C12 is 14.45 (4)°. The angle between the C10—C7 and its symmetry-related bond vector is 25.8 (1)°, and the dihedral angle between the two fluorophenyl rings is 12.15 (10)°. However, the shortest interatomic distance between the fluorophenyl rings is 4.597 (6) Å for C6···C6ⁱ, and there is no intramolecular π — π interaction.

There are intramolecular O2—H2···O3ⁱ hydrogen bonds (Table 2), but no intermolecular hydrogen bonds. This situation contrasts with the crystal structure of the hydroperoxy-biindenylidene compound, which forms a ladder structure through the O—H···O intermolecular hydrogen bonds, accommodating several guest molecules (Tanaka *et al.*, 2003).

Experimental

Compound (I) was obtained in trace amounts as a by-product in the treatment of 3,3'-bis(4-fluorophenyl)-[2,2']biindenyl-1,1'-dione with Zn and ZnCl₂ in aqueous tetrahydrofuran to obtain 3,3'-bis(4-fluorophenyl)-3H,3'H-[2,2']biindenylidene-1,1'-dione (22% yield) and 3,1'-bis(4-fluorophenyl)-1'-hydroperoxy-3'-hydroxy-1',3'-dihydro-3H-[2,2']biindenylidene-1-one (7% yield; Tanaka *et al.*, 2003). The

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Figure 1

A view of the molecular structure of (I), showing 50% probability displacement ellipsoids for non-H atoms. [Symmetry code: (*) $\frac{1}{2} - x, \frac{1}{2} - y, z$.]

reaction mechanism which gave (I) is expected to be similar to that reported by Xu *et al.* (2002). Colourless crystals of (I) were grown from an ethyl acetate solution by slow evaporation (m.p. 482–485 K).

Crystal data

$C_{30}H_{20}F_2O_4$
 $M_r = 482.48$
 Orthorhombic, *Pccn*
 $a = 8.303$ (2) Å
 $b = 12.011$ (3) Å
 $c = 22.868$ (6) Å
 $V = 2280.6$ (10) Å³
 $Z = 4$
 $D_x = 1.405$ Mg m⁻³

Mo $K\alpha$ radiation
 Cell parameters from 20 reflections
 $\theta = 13.0$ – 14.5°
 $\mu = 0.10$ mm⁻¹
 $T = 298$ K
 Plate, colourless
 $0.5 \times 0.4 \times 0.3$ mm

Data collection

Rigaku AFC-7R diffractometer
 ω scans
 Absorption correction: none
 3523 measured reflections
 2622 independent reflections
 1644 reflections with $I > 2\sigma(I)$
 $R_{int} = 0.010$

$\theta_{max} = 27.5^\circ$
 $h = -4 \rightarrow 10$
 $k = -15 \rightarrow 6$
 $l = 0 \rightarrow 29$
 3 standard reflections every 150 reflections
 intensity decay: none

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma F^2] = 0.054$
 $wR(F^2) = 0.190$
 $S = 1.03$
 2622 reflections
 164 parameters
 H-atom parameters not refined

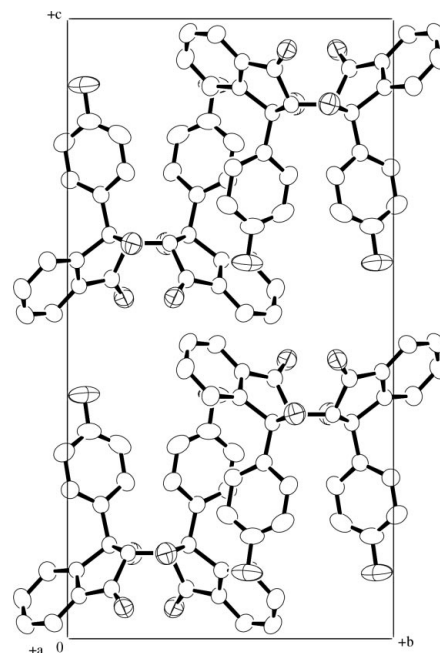
$w = 1/[\sigma^2(F_o^2) + (0.093P)^2 + 1.0454P]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{max} < 0.001$
 $\Delta\rho_{max} = 0.27$ e Å⁻³
 $\Delta\rho_{min} = -0.46$ e Å⁻³

Table 1

Selected geometric parameters (Å, °).

F1—C4	1.359 (3)	C11—C11 ⁱ	1.538 (4)
O2—C10	1.430 (3)	C11—C12	1.533 (3)
O3—C12	1.214 (4)	C12—C13	1.461 (4)
C10—C11	1.563 (4)	C13—C18	1.382 (4)
C10—C18	1.529 (3)		
C10—C11—C11 ⁱ —C10 ⁱ	−154.8 (3)	C10—C11—C11 ⁱ —C12 ⁱ	82.6 (2)

Symmetry code: (i) $\frac{1}{2} - x, \frac{1}{2} - y, z$.


Figure 2

The crystal structure of (I), projected along *a*. H atoms have been omitted for clarity.

Table 2

Hydrogen-bonding geometry (Å, °).

<i>D</i> —H... <i>A</i>	<i>D</i> —H	H... <i>A</i>	<i>D</i> ... <i>A</i>	<i>D</i> —H... <i>A</i>
O2—H2...O3 ⁱ	0.82	1.99	2.766 (3)	158

Symmetry code: (i) $\frac{1}{2} - x, \frac{1}{2} - y, z$.

The hydroxyl H atom was located in a difference synthesis and was allowed to ride on the parent atom. The other H atoms were positioned geometrically and fixed, with $U_{iso}(H) = 1.2U_{eq}(\text{parent atom})$.

Data collection: *WinAFC Diffractometer Control Software* (Rigaku, 1999); cell refinement: *WinAFC Diffractometer Control Software*; data reduction: *TEXSAN* (Molecular Structure Corporation, 2001); program(s) used to solve structure: *SIR92* (Altomare *et al.*, 1994); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEPII* (Johnson, 1976); software used to prepare material for publication: *TEXSAN*.

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